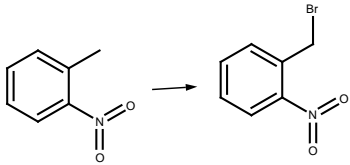
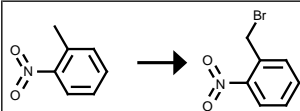


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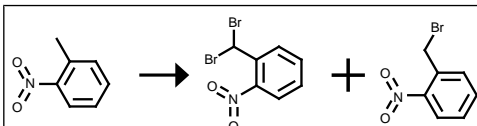
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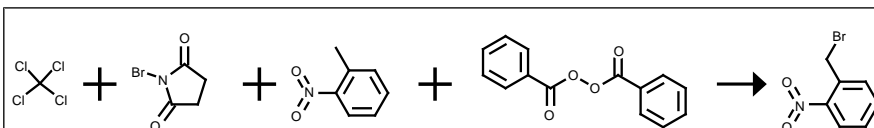
Yield	Conditions & References
84.5 %	<p>Example Name 1</p> <p>O-nitrotoluene (3.4g, 25mmol) is dissolved in 140ml of anhydrous carbon tetrachloride, and benzoyl peroxide (0.290g, 1.2mmol) and N-bromosuccinimide (4.450g, 25mmol) are added. The reaction mixture is stirred, heated at reflux for 4 hours until N-bromosuccinimide is floated up completely. Cooling and then filtrating, washing with cold sodium bicarbonate aqueous solution (2x), and then ice-water (2x), and drying over anhydrous magnesium sulfate overnight. After filtration and concentration, the product is recrystallized with aqueous ethanol to obtain o-nitrobenzyl bromide as a white crystal, with a yield of 84.5percent, m.p. 45-47.deg.C.</p> <p>With N-bromosuccinimide, dibenzoyl peroxide in tetrachloromethane, Time= 4h, Reflux</p> <p>Patent: Jiangsu Goworth Investment Co. Ltd; Shanghai Institute of Pharmaceutical Industry; EP2308868; (2011); (A1) English View in Reaxys</p>
84.5 %	<p>Example Name 1</p> <p>O-nitrotoluene (3.4 g, 25 mmol) is dissolved in 140 ml of anhydrous carbon tetrachloride, and benzoyl peroxide (0.290 g, 1.2 mmol) and N-bromosuccinimide (4.450 g, 25 mmol) are added. The reaction mixture is stirred, heated at reflux for 4 hours until N-bromosuccinimide is floated up completely. Cooling and then filtrating, washing with cold sodium bicarbonate aqueous solution (2.x.), and then ice-water (2.x.), and drying over anhydrous magnesium sulfate overnight. After filtration and concentration, the product is recrystallized with aqueous ethanol to obtain o-nitrobenzyl bromide as a white crystal, with a yield of 84.5percent, m.p. 45-47.deg. C.</p> <p>With N-bromosuccinimide, dibenzoyl peroxide in tetrachloromethane, Time= 4h, Reflux</p> <p>Patent: JIANGSU GOWORTH INVESTMENT CO. LTD; SHANGHAI INSTITUTE OF PHARMACEUTICAL INDUSTRY; US2011/152323; (2011); (A1) English View in Reaxys</p>
81 %	<p>With N-bromosuccinimide, meta-chloroperoxybenzoic acid, Time= 3h, Heating, Irradiation</p> <p>Rigo, Benoit; Dolaine, Regis; Ghammarti, Samira El; Couturier, Daniel; Journal of Heterocyclic Chemistry; vol. 33; nb. 4; (1996); p. 1063 - 1066 View in Reaxys</p>
77 %	<p>With N-bromosuccinimide in tetrachloromethane, Time= 3h, Heating, Irradiation</p> <p>Majumder, P. L.; Sarkar, A. K.; Journal of the Indian Chemical Society; vol. 66; (1989); p. 673 - 680 View in Reaxys</p>
69 %	<p>With N-bromosuccinimide, dibenzoyl peroxide in tetrachloromethane, T= 80 °C</p> <p>Mishra, Jitendra Kumar; Garg, Puja; Dohare, Preeti; Kumar, Ashutosh; Siddiqi, Mohammad Imran; Ray, Madhur; Panda, Gautam; Bioorganic and Medicinal Chemistry Letters; vol. 17; nb. 5; (2007); p. 1326 - 1331 View in Reaxys</p>
46 %	<p>With N-bromosuccinimide, dibenzoyl peroxide in tetrachloromethane, Time= 21h, Heating, Irradiation</p> <p>Collins, David J.; Drygala, Peter F.; Swan, John M.; Australian Journal of Chemistry; vol. 36; nb. 10; (1983); p. 2095 - 2110 View in Reaxys</p>
41 %	<p>With N-bromosuccinimide, meta-chloroperoxybenzoic acid in tetrachloromethane, Time= 9h, Heating</p> <p>Baldwin, Jack E.; Cha, Jin K.; Kruse, Lawrence I.; Tetrahedron; vol. 41; nb. 22; (1985); p. 5241 - 5260 View in Reaxys</p>
30 %	<p>With N-bromosuccinimide, Time= 0.15h, microwave irradiation</p> <p>Goswami, Shyamaprosad; Dey, Swapan; Jana, Subrata; Adak, Avijit Kumar; Chemistry Letters; vol. 33; nb. 7; (2004); p. 916 - 917</p>

<p>29 %</p>	<p>View in Reaxys</p> <p>Example Name 1.1 A mixture of l-methyl-2-nitrobenzene (20 g, 144.53 mmol), NBS (28.4 g, 158.85 mmol), and AIBN (0.2 g) in CCl₄ (150 mL) was refluxed for 3h, then cooled to RT, filtered, and concentrated under vacuum. The residue was partitioned between EtOAc and H₂O and the aqueous phase was extracted with EtOAc (3x300ml). The organics were combined, dried over Na₂SO₄, and concentrated under vacuum. The residue was purified by column chromatography on silica gel (eluting with 1 :20 <n="54"/>Attorney Docket No. KO 152-401 -PCEtOAc/PE) to afford 10 g (29percent) of l-(bromo-methyl)-2-nitrobenzene as a white solid. ¹HNMR (300MHz, DMSOd₆) δ 8.08 (s, 1H), 7.75 (m, 2H), 7.60 (m, 1H), 4.94 (s, 2H)</p> <p>With N-bromosuccinimide, 2,2'-azo-bisisobutyronitrile in tetrachloromethane, Time= 3h, Heating / reflux</p> <p>Patent; KALYPSSYS, INC.; WO2009/29625; (2009); (A1) English View in Reaxys</p>
	<p>With tetrachloromethane, N-bromosuccinimide, dibenzoyl peroxide</p> <p>Kornblum; Iffland; Journal of the American Chemical Society; vol. 71; (1949); p. 2137,2138,2140 View in Reaxys</p>
	<p>With (R)-bromo alcohol, dibenzoyl peroxide</p> <p>Koeroesi,J.; Monatshefte fuer Chemie; vol. 100; (1969); p. 1222 - 1232 View in Reaxys</p>
	<p>With N-bromosuccinimide</p> <p>Huppertz,J.L.; Australian Journal of Chemistry; vol. 26; (1973); p. 1307 - 1318 View in Reaxys Matsumoto,I.; Chemical and Pharmaceutical Bulletin; vol. 15; (1967); p. 1990 - 1995 View in Reaxys</p>
	<p>With bromine, dibenzoyl peroxide in tetrachloromethane, Irradiation</p> <p>Celnik,K.; Jankowski,Z.; Polish Journal of Chemistry; vol. 52; (1978); p. 947 - 951 View in Reaxys</p>
	<p>With N-bromosuccinimide, meta-chloroperoxybenzoic acid in tetrachloromethane, Time= 5h, Heating</p> <p>Boyer, S. K.; Fitchett, G.; Wasley, J. W. F.; Zaunius, G.; Journal of Heterocyclic Chemistry; vol. 21; (1984); p. 833 - 836 View in Reaxys</p>
	<p>With N-bromosuccinimide, meta-chloroperoxybenzoic acid in tetrachloromethane</p> <p>Corre, M. Le; Hercouet, A.; Stanc, Y. Le; Baron, H. Le; Tetrahedron; vol. 41; nb. 22; (1985); p. 5313 - 5320 View in Reaxys</p>
	<p>With N-bromosuccinimide, meta-chloroperoxybenzoic acid</p> <p>Sikkar, Rein; Martinson, Per; Acta Chemica Scandinavica, Series B: Organic Chemistry and Biochemistry; vol. 34; nb. 8; (1980); p. 551 - 558 View in Reaxys</p>
	<p>With N-bromosuccinimide, meta-chloroperoxybenzoic acid in tetrachloromethane, Time= 6h, Heating</p> <p>Stambach, JF; Kanmacher, I; Jung, L; Schott, C; Heitz, C; Stoclet, JC; European Journal of Medicinal Chemistry; vol. 28; nb. 5; (1993); p. 427 - 432 View in Reaxys</p>
	<p>With N-bromosuccinimide</p> <p>Berg, E. M. M. van den; Baldew, A. U.; Goede, A. T. J. W. de; Raap, J.; Lugtenburg, J.; Recueil des Travaux Chimiques des Pays-Bas; vol. 107; nb. 2; (1988); p. 73 - 81 View in Reaxys</p>

	<p>With N-bromosuccinimide, dibenzoyl peroxide in tetrachloromethane, Heating</p> <p>Jan, Thierry; Dupas, Beatrice; Floner, Didier; Moinet, Claude; Tetrahedron Letters; vol. 43; nb. 34; (2002); p. 5949 - 5952</p> <p>View in Reaxys</p>
	<p>With N-bromosuccinimide, dibenzoyl peroxide in tetrachloromethane, Time= 5h, Heating</p> <p>Murata, Shigeru; Tsubone, Yasuhiro; Kawai, Reina; Eguchi, Daisuke; Tomioka, Hideo; Journal of Physical Organic Chemistry; vol. 18; nb. 1; (2005); p. 9 - 20</p> <p>View in Reaxys</p>
	<p>With N-bromosuccinimide, dibenzoyl peroxide in tetrachloromethane, Time= 12h, Heating</p> <p>Makhija, Mahindra T.; Kasliwal, Rajesh T.; Kulkarni, Vithal M.; Neamati, Nouri; Bioorganic & Medicinal Chemistry; vol. 12; nb. 9; (2004); p. 2317 - 2334</p> <p>View in Reaxys</p>
	<p>With N-bromosuccinimide, Time= 16h, T= 36 °C , Irradiation, neat (no solvent)</p> <p>Jereb, Marjan; Zupan, Marko; Stavber, Stojan; Helvetica Chimica Acta; vol. 92; nb. 3; (2009); p. 555 - 566</p> <p>View in Reaxys</p>
	<p>Example Title General procedure for the bromination of methylarenes.</p> <p>General procedure: To a solution of the methylarene derivative (1.0 eq.) in C6H6 is added N-bromosuccinimide (1.05 eq.), azobisisobutyronitrile (0.25 eq.) and Br2 (0.25 mL). The solution was heated to reflux for 24h. Upon completion, the reaction mixture was cooled and washed with H2O (3X). The organic layer was washed successively with sat'd. aq. Na2S2O3 (2X), sat'd. aq. NaCl (1X), dried over MgSO4 and concentrated.</p> <p>With N-bromosuccinimide, 2,2'-azo-bisisobutyronitrile, bromine in benzene, Time= 24h, Reflux</p> <p>Motto, John M.; Montemayer, Laura K.; Sheepwash, Erin E.; Schwan, Adrian L.; Castillo, Alvaro; Greer, Alexander; Tetrahedron; vol. 67; nb. 5; (2011); p. 1002 - 1010</p> <p>View in Reaxys</p>


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Yield	Conditions & References
41 % Chromat., 52 % Chromat.	<p>With bromine in tetrachloromethane, Heating, Irradiation</p> <p>Ong-Lee, Avelina; Sylvester, Leo; Wasley, Jan W. F.; Journal of Heterocyclic Chemistry; vol. 20; (1983); p. 1565 - 1569</p> <p>View in Reaxys</p>
52 % Chromat., 41 % Chromat.	<p>With bromine in tetrachloromethane, Heating, Irradiation</p> <p>Ong-Lee, Avelina; Sylvester, Leo; Wasley, Jan W. F.; Journal of Heterocyclic Chemistry; vol. 20; (1983); p. 1565 - 1569</p> <p>View in Reaxys</p>


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Yield	Conditions & References
	<p>Kornblum; Iffland; Journal of the American Chemical Society; vol. 71; (1949); p. 2137,2138,2140</p> <p>View in Reaxys</p>

