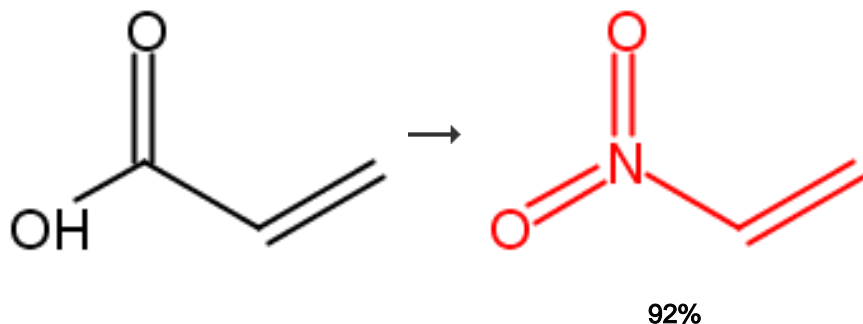


**1. Single Step**[Overview](#)**Steps/Stages**

- 1.1 C:15025-74-8, S:HOCH<sub>2</sub>CH<sub>2</sub>OH polymer, S:ClCH<sub>2</sub>CH<sub>2</sub>Cl, S:MeCN, 1.5 h, reflux; cooled
- 1.2 R:Disodium carbonate, S:H<sub>2</sub>O

**Notes**

optimization study, PEG-300 used as solvent, optimized on solvent (PEG), Hunsdiecker reaction, Reactants: 1, Reagents: 1, Catalysts: 1, Solvents: 4, Steps: 1, Stages: 2, Most stages in any one step: 2

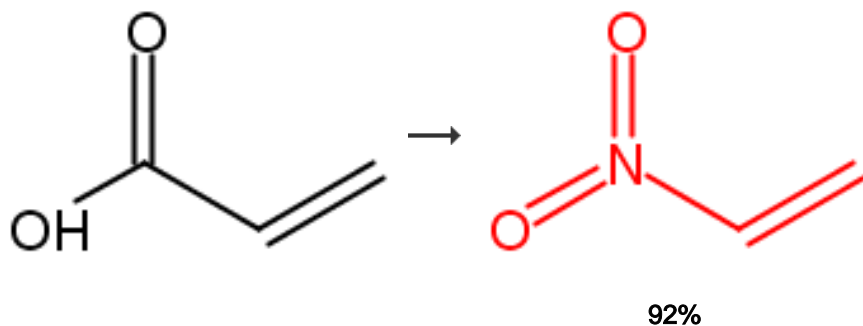
**References**

[Polyethylene glycol mediated kinetic study of nitro decarboxylation of  \$\alpha,\beta\$ -unsaturated acids by Blau's Fe\(III\) Phen complex](#)

By Ramesh, K. et al

From Journal of Chemistry, , 703271/1-703271/11, 11 pp.; 2013

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**2. Single Step**[Overview](#)**Steps/Stages****Notes**

- 1.1 R:Fe(NO<sub>3</sub>)<sub>3</sub>, R:2,2'-(C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>, S:MeCN, rt  
 1.2 S:HOCH<sub>2</sub>CH<sub>2</sub>OH polymer, 1.5 h, rt  
 1.3 R:Disodium carbonate, S:H<sub>2</sub>O, rt

green chemistry-process simplification, in-situ generated reagent [(Fe(III)-Bipy) complex] formed in stage 1, see experimental details, kinetics studied, mechanism studied, PEG 300 used as solvent in stage 2, optimization study in stage 2, optimized on type of PEG, green chemistry-solvent (PEG), Nitro Hunsdiecker reaction, alternatively reaction carried out in absence of PEG in stage 2 decreased yield and increased reaction time, solvent recyclable (PEG), Reactants: 1, Reagents: 3, Solvents: 3, Steps: 1, Stages: 3, Most stages in any one step: 3

### References

[Polyethylene Glycol-Mediated Kinetic Study of Nitrodecarboxylation of  \$\alpha,\beta\$ -Unsaturated Acids by Blau Fe\(III\) Bipy Complex](#)

By Ramesh, K. et al

From International Journal of Chemical Kinetics, 46(2), 126-137; 2014

### Reaction Protocol

#### Procedure

1. Suspend the AA (0.001 mol) dissolved in MeCN and 0.001 moles of Blau's[Fe (III)-Bipy] complex in a three-necked round-bottomed flask, equipped with a mechanical stirrer and condenser.
2. Reflux the reaction mixture for 1.5 hours at room temperature until completion of the reaction.

[View more...](#)

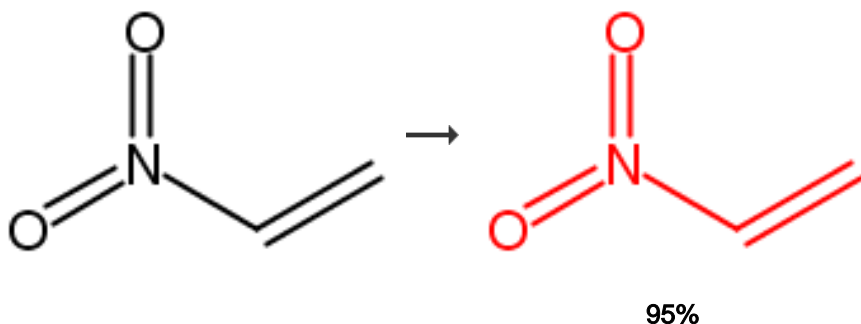
#### Available Experimental Data

<sup>1</sup>H NMR, Mass Spec

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### 3. Single Step



[Overview](#)

[Steps/Stages](#)

[Notes](#)

1.1 R:NaOEt, S:EtOH, 14 h, rt

Reactants: 1, Reagents: 1, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[Silicon-catalyzed conversion of nitro compounds into ketones and poly\(1,3-diketones\)](#)

By Hwu, Jih Ru et al

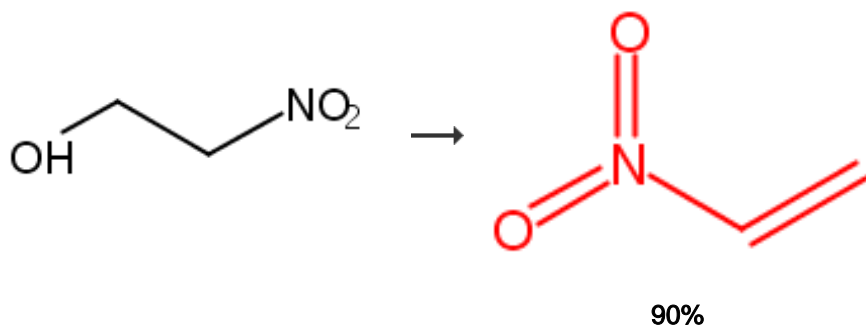
From *Synthesis*, (19), 3305-3308; 2006

#### Experimental Procedure

**Poly(nitroethylene) (6)<sup>15b</sup>.** To a stirred soln of nitroethene (**4**; 5.03 g, 68.9 mmol, 1.0 equiv) in EtOH (690 mL) was added NaOEt (234 mg, 3.44 mmol, 0.050 equiv). The mixture was stirred at r.t. for 14 h, and after this time a white solid was obtained. The solid was purified by washing with H<sub>2</sub>O (1 x), MeOH (2 x), Et<sub>2</sub>O (2 x) and then dried under reduced pressure over P<sub>2</sub>O<sub>5</sub> to give polymer **6**. Yield 4.78 g, 95%. soluble in DMSO, THF, and DMF; MW 7.12 x 10<sup>4</sup>. IR (KBr): 3000 (m), 2907 (m), 1559 (br s, N-O), 1436 (m), 1363 (br m, N-O), 849 (s, =CH) cm<sup>-1</sup>. <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>): δ = 2.49-2.65 (br d, 2 H, CH<sub>2</sub>), 4.59- 4.75 (br m, 1 H, CHNO<sub>2</sub>). Anal. Calcd for (C<sub>2</sub>H<sub>3</sub>NO<sub>2</sub>)<sub>n</sub>: C, 32.88; H, 4.14; N, 19.17. Found: C, 33.01; H, 4.15; N, 19.22.

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#### 4. Single Step



#### Overview

#### Steps/Stages

1.1 R:Phthalic anhydride, 110-140°C, 60 mmHg

#### Notes

Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[Aminoethylation of aldehydes in synthesis of γ-amino acids](#)

By Gellman, Samuel Helmer et al

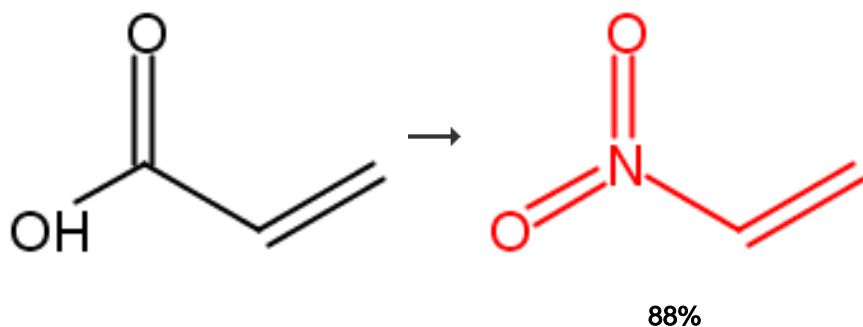
From U.S. Pat. Appl. Publ., 20090264676, 22 Oct 2009

#### Experimental Procedure

**Nitroethylene** This compound was prepared using a modification of a literature procedure (see Kunetsky et al., *Tetrahedron Lett.* 2005, 46, 5203; Ranganathan et al, *Tetrahedron Lett.* 1987, 28, 2893). 2-Nitroethanol (100 g) and phthalic anhydride (210 g) were mixed in a 500 ml round bottom flask equipped with a magnetic stir bar. The flask was then equipped with a vacuum distillation setup with a fractional distillation column and a -78°C. bath-cooled receiver. The apparatus was evacuated to about 60 mm Hg, and the oil bath was heated to and maintained at 110-140°C. The starting materials turned to a homogeneous solution (solid material may exist depending on the temperature), and the distillate was collected until the distillation ceased to give a pale yellow solid at -78°C. The solid, containing a mixture of nitroethylene and water (~90 g, 90% yield), was warmed in an icewater bath to give a pale yellow heterogeneous mixture. The mixture was mixed with toluene, dried over anhydrous CaCl<sub>2</sub> (anhydrous), and filtered through a pad of anhydrous CaCl<sub>2</sub>. The filtrate was collected as a pale yellow stock solution of nitroethylene in toluene, and stored at -10°C. for future use. The concentration of nitroethylene of the stock solution can be estimated via <sup>1</sup>H NMR analysis in benzene-d<sub>6</sub> (with toluene as internal standard). Concentration estimated from the <sup>1</sup>H NMR analysis agreed with that calculated from mass of the crude nitroethylene product (the impurity is water) and toluene used in preparation of the stock solution. Nitroethylene as a solution in dry benzene was found to be stable (no change in NMR) for at least 6 months when stored in a refrigerator. Nitroethylene, yield (~90 g, 90% yield). <sup>1</sup>H NMR of stock solution in toluene (300 MHz, CDCl<sub>3</sub>) δ 7.11 (dd, J=7.2,15 Hz), 6.62 (dd, J=14.7,2.1 Hz), 5.86 (br d, J=7 Hz); <sup>13</sup>C NMR of stock solution in toluene (75 MHz, CDCl<sub>3</sub>) δ 145.56, 122.36.

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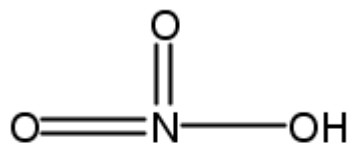
### 5. Single Step



#### Overview

#### Steps/Stages

1.1 R:



• 1/3 Al

C:HOCH<sub>2</sub>CH<sub>2</sub>OH polymer, S:MeCN, 1.5 h

#### Notes

regioselective, optimized on metal nitrate, type of PEG and method, PEG-600 used as catalyst, Hunsdiecker-Borodin reaction, green chem.-catalyst, optimization study, Reactants: 1, Reagents: 1, Catalysts: 1, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

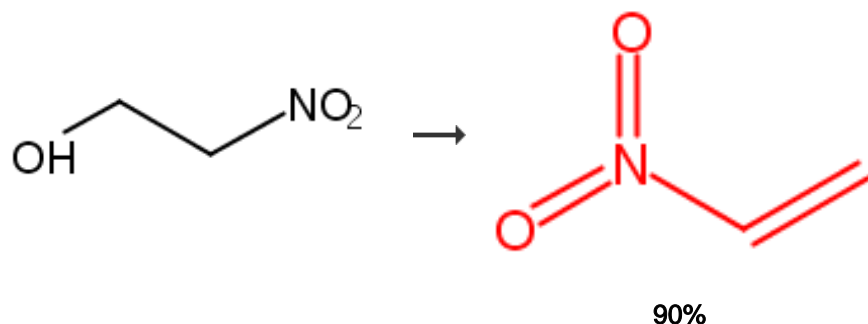
[Poly ethylene glycols as efficient media for the synthesis of β-nitro styrenes from α, β-unsaturated carboxylic acids and metal nitrates under conventional and non-conventional conditions](#)

By Rajanna, Kamatala Chinna et al

From *Green and Sustainable Chemistry*, 1(4), 132-148; 2011

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### 6. Single Step



## Overview

### Steps/Stages

1.1 R:Phthalic anhydride, rt → 110°C; 110-140°C

### Notes

low pressure, Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

### References

[Enantioselective organocatalytic Michael addition of aldehydes to nitroethylene: efficient access to  \$\gamma\$ -amino acids](#)

By Chi, Yonggui et al

From Journal of the American Chemical Society, 130(17), 5608-5609; 2008

## Experimental Procedure

**Nitroethylene:** This compound was prepared using a modification of a literature procedure. 7, 8 2-Nitroethanol (100 g) and phthalic anhydride (210 g) were mixed in a 500 ml round bottom flask equipped with a magnetic stir bar. The flask was then equipped with a vacuum distillation setup with a fractional distillation column and a -78 °C bath-cooled receiver. The apparatus was evacuated to about 60 mmHg, and the oil bath was heated to and maintained at 110-140 °C. The starting materials turned to a homogeneous solution (solid material may exist depending on the temperature), and the distillate was collected until the distillation ceased to give a pale yellow solid at -78 °C. The solid, containing a mixture of nitroethylene and water (~90g, 90% yield), was warmed in an ice-water bath to give a pale yellow heterogeneous mixture. The mixture was mixed with toluene, dried over anhydrous CaCl<sub>2</sub> (anhydrous), and filtered through a pad of anhydrous CaCl<sub>2</sub>. The filtrate was collected as a pale yellow stock solution of nitroethylene in toluene, and stored at -10 °C for future use. The concentration of nitroethylene of the stock solution can be estimated via <sup>1</sup>H NMR analysis in benzene-d<sub>6</sub> (with toluene as internal standard). Concentration estimated from the <sup>1</sup>H NMR analysis agreed with that calculated from mass of the crude nitroethylene product (the impurity is water) and toluene used in preparation of the stock solution. Nitroethylene as a solution in dry benzene was found to be stable (no change in NMR) for at least 6 months when stored in a refrigerator. **Nitroethylene:** Yield (~90g, 90%). <sup>1</sup>H NMR of stock solution in toluene (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.11 (dd,  $J$  = 7.2, 15 Hz), 6.62 (dd,  $J$  = 14.7, 2.1 Hz), 5.86 (br d,  $J$  = 7 Hz); <sup>13</sup>C NMR of stock solution in toluene (75 MHz, CDCl<sub>3</sub>)  $\delta$  145.56, 122.36.

## Reaction Protocol

### Procedure

1. Mix 7, 8 2-nitroethanol (100 g) and phthalic anhydride (210 g) in a 500 ml round bottomed flask equipped with a magnetic stir bar.
2. Equip the flask with a vacuum distillation setup with a fractional distillation column.

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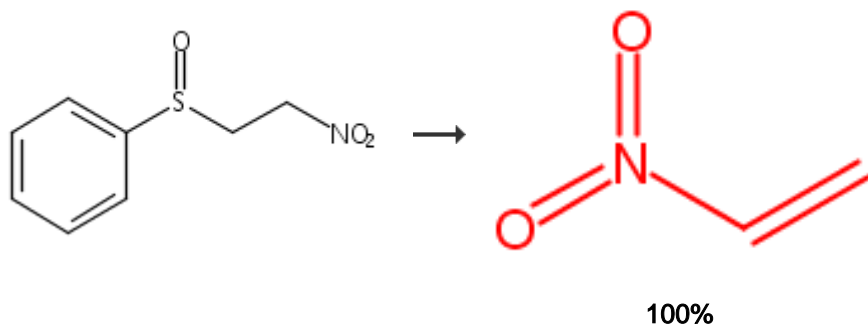
### Available Experimental Data

<sup>1</sup>H NMR, <sup>13</sup>C NMR

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### 7. Single Step



#### Overview

#### Steps/Stages

1.1 S: Benzene

#### Notes

Reactants: 1, Solvents: 1, Steps: 1, Stages: 1,  
Most stages in any one step: 1

#### References

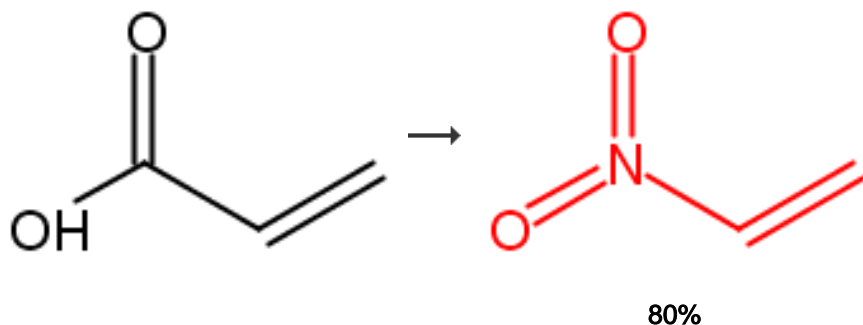
[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987

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### 8. Single Step



#### Overview

#### Steps/Stages

#### Notes

1.1 R:t-BuONO, S:MeCN, 4-5 min, 100°C, 3 bar

microwave irradiation, silica gel used, alternative conditions (conventional, sonication) gave lower yield, small amount of solvent used (solvent-free conditions-author emphasis), Reactants: 1, Reagents: 1, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[Ultrasonic and microwave-assisted synthesis of  \$\beta\$ -nitro styrenes and nitro phenols with tertiary butyl nitrite under acid-free conditions](#)

By Kumar, M. Satish et al

From Synthetic Communications, 43(19), 2672-2677; 2013

#### Experimental Procedure

General/Typical Procedure: **General Procedure for Microwave-Assisted Synthesis of Nitro Compounds Under Solvent-Free Conditions** Organic substrate (cinnamic acids, phenols), TBN and solvent (small amount of acetonitrile) were mixed with silica gel and heated in a controlled microwave synthesizer (Biotage Initiator+SP Wave model 0.200W at 2.45 GHz, capped at 60W during steady state) for 4-5 min (attained temperature of 100 °C and 3 bar pressure). Progress of the reaction was monitored by TLC. After completion, the reaction mixture was further processed for the isolation of product as described previously. **1-Nitro Ethene**, yield 80%. Mp 98-102 °C (98.5 °C) <sup>1</sup>HNMR (300MHz, CDCl<sub>3</sub>):  $\delta$  5.92 (dd, 1H, H<sub>b</sub>, J=11.3 Hz, J=1.5Hz); 6.65 (dd, 1H, H<sub>a</sub>, J=15.5 Hz, J=11.3 Hz); 7.25 (dd, 1H, H<sub>c</sub>, J=15.5Hz, J=1.5 Hz), m/z = 73.

#### Reaction Protocol

##### Procedure

1. Mix acid derivative, tert-butyl nitrite and acetonitrile with silica gel.
2. Heat the mixture in a controlled microwave synthesizer (Biotage Initiator+SP Wave model 0.200W at 2.45 GHz, capped at 60W during steady state) for 4-5 minutes (attained temperature of 100 °C and 3 bar pressure).

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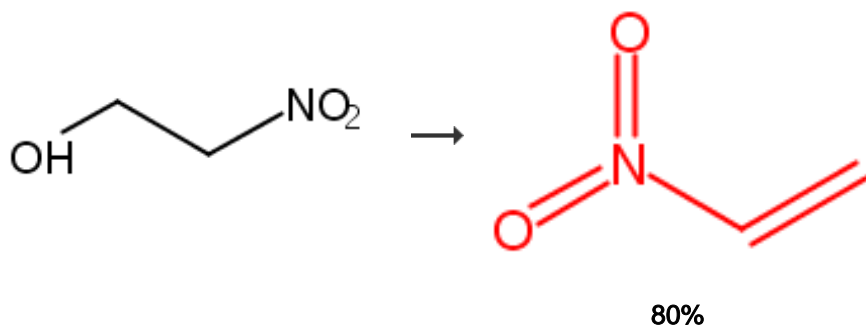
#### Available Experimental Data

<sup>1</sup>H NMR, Mass Spec, MP

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#### 9. Single Step



[Overview](#)

[Steps/Stages](#)

[Notes](#)

1.1 R:Phthalic anhydride, 140-150°C, 80 mmHg; 175-180°C

thermal, no solvent, low pressure, Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

### References

[Preparation of N-pyridinyl carboxamide derivatives as modulators of ATP-binding cassette transporters](#)

By Hadida-Ruah, Sara et al

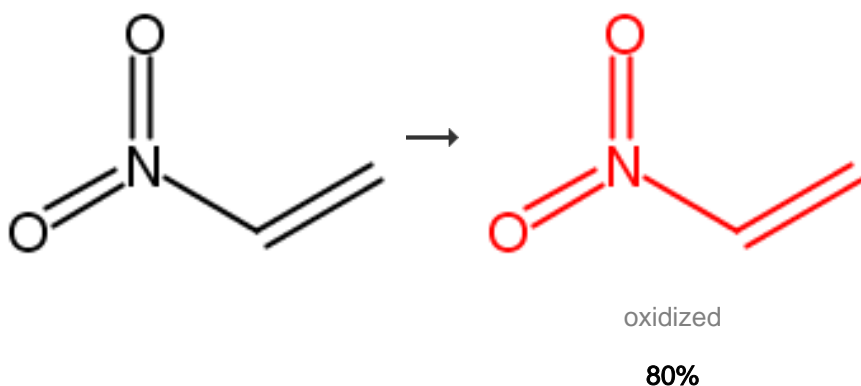
From U.S. Pat. Appl. Publ., 20080019915, 24 Jan 2008

### Experimental Procedure

Nitroethylene 2-Nitroethanol (3.5 g, 39 mmol) and sublimed phthalic anhydride (7.5 g, 58 mmol) were mixed in a distillation unit with a short fractional column and an ice-cooled receiver. The apparatus was evacuated to 80 mm of Hg, and the bath temperature was maintained at 140-150° C. until the mixture was homogeneous. The temperature was increased and held at 175-180° C. until distillation ceased. The distillate was dried over anhydrous CaCl<sub>2</sub> to give nitroethylene (2.3 g, 80%). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz) δ 7.16-7.10 (m, 1H), 6.66-6.62 (m, 1H), 5.91-5.90 (m, 1H).

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### 10. Single Step



### Overview

#### Steps/Stages

- 1.1 R:NaOEt, S:Dioxane, 24 h, rt
- 1.2 R:KH, S:Dioxane, 1.5 h, 10°C
- 1.3 R:Me<sub>3</sub>SiCl, 2 h, rt; 24 h, reflux

#### Notes

Reactants: 1, Reagents: 3, Solvents: 1, Steps: 1, Stages: 3, Most stages in any one step: 3

### References

[Silicon-catalyzed conversion of nitro compounds into ketones and poly\(1,3-diketones\)](#)

By Hwu, Jih Ru et al

From Synthesis, (19), 3305-3308; 2006

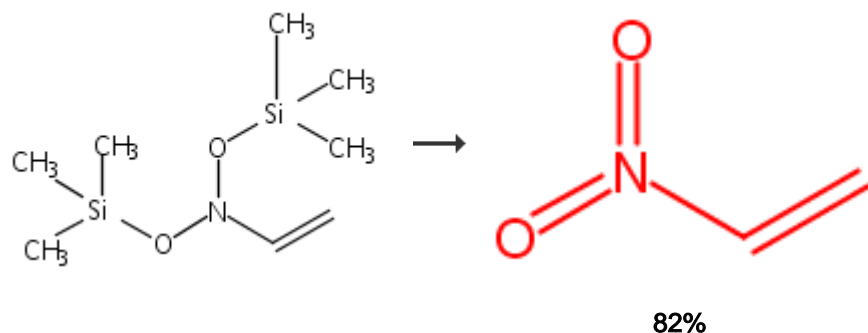
### Experimental Procedure

**Poly(oxoethylene) (2); Method B in One Flask.** NaOEt (149 mg, 0.050 equiv) was added to a soln of nitroethene (**4**; 3.20 g, 43.8 mmol, 1.0 equiv) in 1,4-dioxane (439 mL) and the mixture was stirred at r.t. for 24 h. Then a 1,4-dioxane soln of KH (1.94 g, 1.0 equiv) was added to the mixture and stirring was continued at 10 °C for 1.5 h. Me<sub>3</sub>SiCl (477 mg, 4.39 mmol, 0.10 equiv) was injected into the soln and this was stirred at r.t. for 2.0 h and then heated at reflux for 24 h. The cooled mixture was diluted with hexanes and a brown solid was precipitated. The solid was washed with hexanes and CCl<sub>4</sub> and then was dried under reduced pressure over P<sub>2</sub>O<sub>5</sub> to give polymer **2**. Yield 1.47 g, 80%.



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### 11. Single Step



#### Overview

#### Steps/Stages

1.1 R:Bu<sub>4</sub>N<sup>+</sup> •OAc, R:Br<sub>2</sub>, S:CH<sub>2</sub>Cl<sub>2</sub>, -78°C; 15 min, -78°C

#### Notes

Reactants: 1, Reagents: 2, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

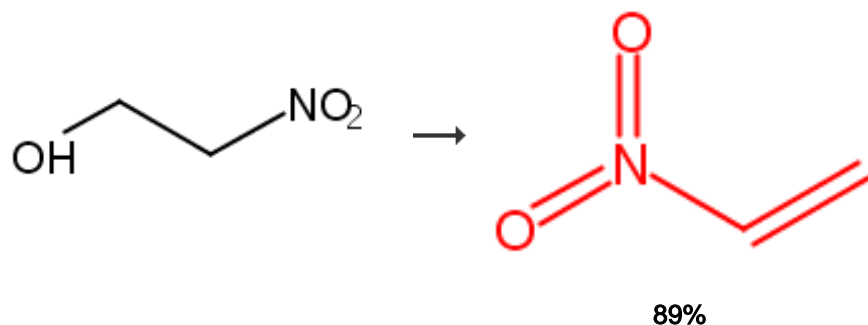
[Novel synthesis of α-nitroalkenes from nitroalkanes via halogenation of intermediate N,N-bis\(silyloxy\)enamines](#)

By Kunetsky, Roman A. et al

From Tetrahedron Letters, 46(31), 5203-5205; 2005

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### 12. Single Step



#### Overview

#### Steps/Stages

1.1 R:Phthalic anhydride

#### Notes

Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[Synthesis of Tricyclic Nitrogen-Containing Systems](#)

By Grohmann, Franz

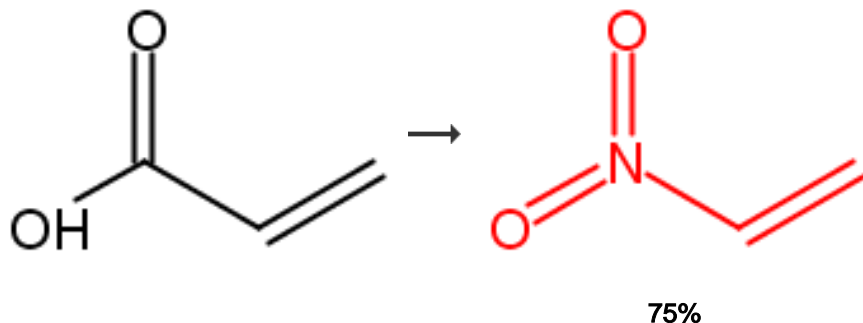
From null, , No pp.; 1991

#### Experimental Procedure

Nitroethene (L5): 5.0 g (55 mmol) 2-nitroethanol and 10.4 g (84 mmol) of phthalic acid anhydride were heated at 100 mbar in a distillation apparatus to 140-150 degC, until the solution was clear. Then the bath temperature was increased to 180 degC, while the nitroethene was distilled collected in a receiver (cooled to -30 degC). Yield: 3.6 g of pale yellow liquid (89 % of theory). bp 30-34 degC/110 mbar.

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### 13. Single Step



#### Overview

#### Steps/Stages

- 1.1 R:Cl(O=)CC(=O)Cl, R:DMF, S:MeCN, -5°C
- 1.2 R:KNO<sub>3</sub>, S:MeCN, 40-60 min, rt

#### Notes

optimization study, optimized on reagent and methods (conventional and microwave), selective nitration, in-situ generated reagent (iminium salt) (stage 1), ultrasound (stage 2), Reactants: 1, Reagents: 3, Solvents: 1, Steps: 1, Stages: 2, Most stages in any one step: 2

#### References

[Oxalyl chloride/DMF as an Efficient Reagent for Nitration of Aromatic Compounds and Nitro Decarboxylation of Cinnamic Acids in Presence of KNO<sub>3</sub> or NaNO<sub>2</sub> Under Conventional and Nonconventional Conditions](#)

By Kumar, M. Satish et al

From *Synthesis and Reactivity in Inorganic, Metal-Organic, and Nano-Metal Chemistry*, 43(8), 977-983; 2013

#### Experimental Procedure

General/Typical Procedure: **General Procedure for Synthesis of Nitro Arenes and β-Nitro Styrenes Using (COCl)<sub>2</sub>+DMF Iminium Salt (Under Sonication)** Organic substrate, KNO<sub>3</sub> (or NaNO<sub>2</sub>), [(COCl)<sub>2</sub>+DMF] iminium salt and solvent (MeCN) were taken in a clean conical flask at room temperature and immersed in a sonicator and progress of the reaction monitored by TLC. After completion, the reaction mixture is further processed for the isolation of product as detailed in earlier section. **β-Nitro styrene 1-Nitro Ethene** Yield 75%. δ5.92 (dd, 1H, Hb, J=11.3 Hz, J=1.5Hz); 6.65 (dd, 1H, Ha, J=15.5 Hz, J=11.3 Hz); 7.25 (dd, 1H, Hc, J=15.5Hz, J=1.5 Hz). m/z -73

#### Reaction Protocol

##### Procedure

1. Take acrylic acid, KNO<sub>3</sub>, [(COCl)<sub>2</sub>+DMF] iminium salt and solvent (MeCN) in a clean conical flask at room temperature.
2. Immerse the mixture in a sonicator.

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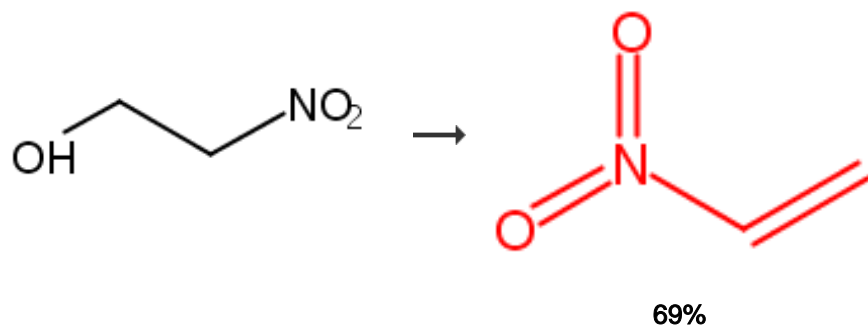
#### Available Experimental Data

<sup>1</sup>H NMR, Mass Spec

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MethodsNow](#)

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#### 14. Single Step



[Overview](#)

#### Steps/Stages

- 1.1 R:Phthalic anhydride, cooled; 130°C; 1 h, 130°C; 2 h, 130°C → 180°C

#### Notes

Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

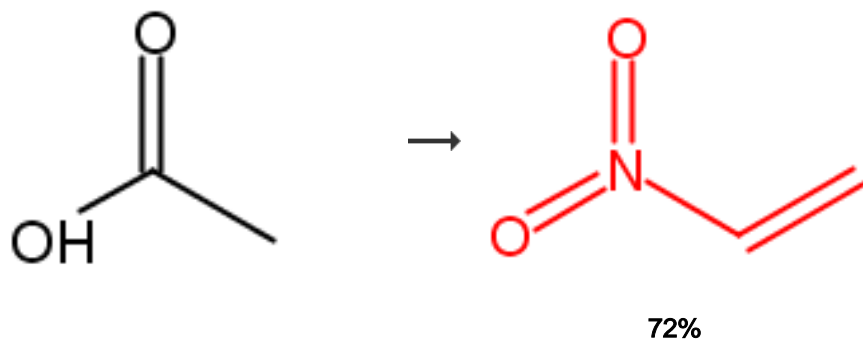
[Preparation of arylcyclopentenones as herbicides](#)

By Hachisu, Shuji et al

From PCT Int. Appl., 2016062585, 28 Apr 2016

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#### 15. Single Step



[Overview](#)

#### Steps/Stages

#### Notes

1.1 R:KNO<sub>3</sub>, R:DMF, R:POCl<sub>3</sub>, rt

green chemistry-process simplification, Vilsmeier-Haack conditions used, no solvent, microwave irradiation (300 W), mechanism studied, optimization study, optimized on reaction conditions, Reactants: 1, Reagents: 3, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

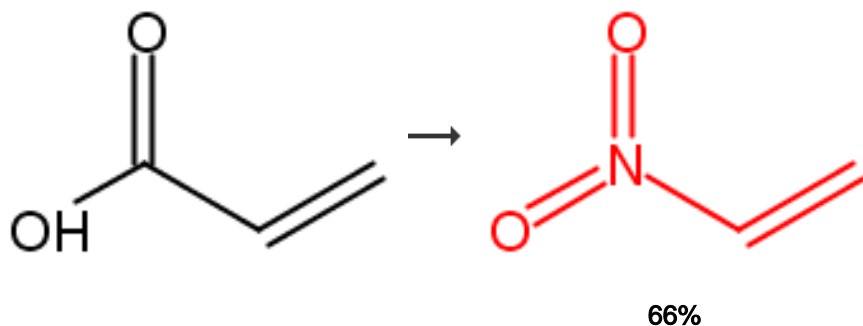
[Iminium salt-mediated nitro decarboxylation of  \$\alpha\$ ,  \$\beta\$ -unsaturated acids for the synthesis of  \$\beta\$ -nitro styrenes under non-conventional conditions](#)

By Mukka, Satish Kumar et al

From Organic Communications, 5(2), 42-49; 2012

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#### 16. Single Step



#### Overview

#### Steps/Stages

1.1 R:HNO<sub>3</sub>, S:MeCN, 180 s, rt

#### Notes

stereoselective, microwave irradiation, green chemistry-catalyst, reusable catalyst, zeolite Y used as catalyst, alternatively conventional heating and sonication method shown, Reactants: 1, Reagents: 1, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

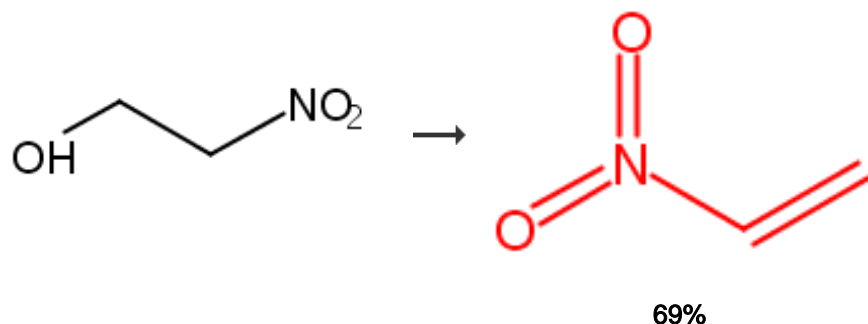
[Zeolite Y-assisted nitration of aromatic and heterocyclic compounds and decarboxylative nitration of  \$\alpha\$ , \$\beta\$ -unsaturated acids under non-conventional conditions](#)

By Sudhakar Chary, V. et al

From Catalysis Science & Technology, 6(5), 1430-1434; 2016

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#### 17. Single Step



### Overview

#### Steps/Stages

- 1.1 R:Phthalic anhydride, -78°C → 130°C; 1 h, 130°C; 2 h, 130°C → 180°C

#### Notes

Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

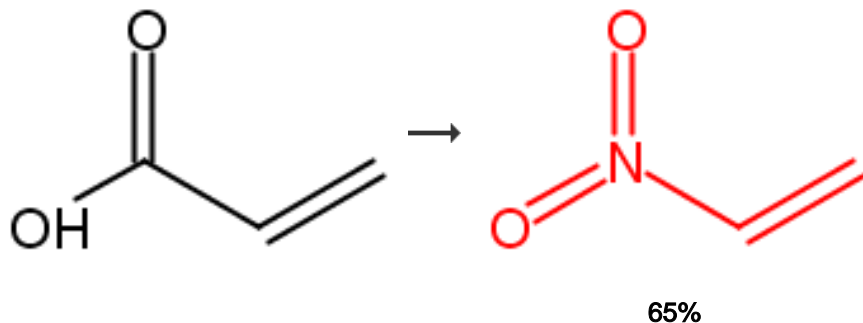
[Preparation of herbicidally active 2-\(substituted-phenyl\)-cyclopentane-1,3-dione compounds and derivatives thereof](#)

By Avery, Alaric James et al

From PCT Int. Appl., 2014170413, 23 Oct 2014

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#### 18. Single Step



### Overview

#### Steps/Stages

- 1.1 R:Isocyanuric chloride, R:DMF, R:NaNO<sub>2</sub>, S:CH<sub>2</sub>Cl<sub>2</sub>, 85 min, rt

#### Notes

ultrasound, green chemistry, alternative reaction conditions gave lower yield (Conventional method), other solvent may also used (acetonitrile), Reactants: 1, Reagents: 3, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[Ultrasonically Assisted Rate Enhancements in Trichloroisocyanuric Acid/DMF/NaNO<sub>2</sub> Triggered Nitration of Aromatic Compounds and Decarboxylative Nitration of  \$\alpha,\beta\$ -Unsaturated Acids](#)

By Satish Kumar, Mukka et al

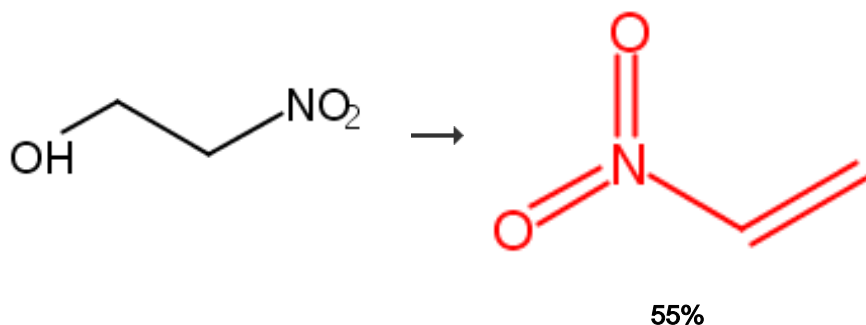
From Synthetic Communications, 45(19), 2251-2258; 2015

## Experimental Procedure

General/Typical Procedure: **Ultrasonically Assisted Synthesis of Nitro Compounds Using TCICA=DMF Reagent with Sodium Nitrite** The reaction mixture containing centimolar molar organic substrate, [TCICA=DMF] reagent, and sodium nitrite were added to the CH<sub>2</sub>Cl<sub>2</sub> or acetonitrile in a clean round-bottomed flask clamped in an ultrasonic bath with a frequency of 33 kHz and 100 W electric power rating and stirred for about 1.0 to 1.5 h. After completion, the reaction mixture was treated with NaHCO<sub>3</sub> solution, followed by the addition of ethyl acetate. The separation and purification procedure is almost the same as the previous procedure. **1-Nitro Ethene**. <sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>): δ 5.92 (dd, 1H, H<sub>b</sub>, J=11.3 Hz, J = 1.5Hz), 6.65 (dd, 1H, H<sub>a</sub>, J=15.5 Hz, J=11.3 Hz), 7.25 (dd, 1H, H<sub>c</sub>, J=15.5Hz, J=1.5 Hz); m/z = 73.

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## 19. Single Step



## Overview

## Steps/Stages

- 1.1 R:Phthalic anhydride, -20 - -5°C, 0.1 MPa; -5°C → 150°C; 1 h, 150°C; 2 h, 150°C → 180°C

## Notes

thermal, Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

## References

[Asymmetric dearomatization of β-naphthols through a bifunctional-thiourea-catalyzed Michael reaction](#)

By Wang, Shou-Guo et al

From *Angewandte Chemie, International Edition*, 54(49), 14929-14932; 2015

## Reaction Protocol

- Procedure**
1. Add nitroethanol (13.65 g) and phthalic anhydride (28.86 g, 0.195 mol) to a flask equipped with distillation apparatus.
  2. Evacuate the flask to -0.1 mPa.

[View more...](#)

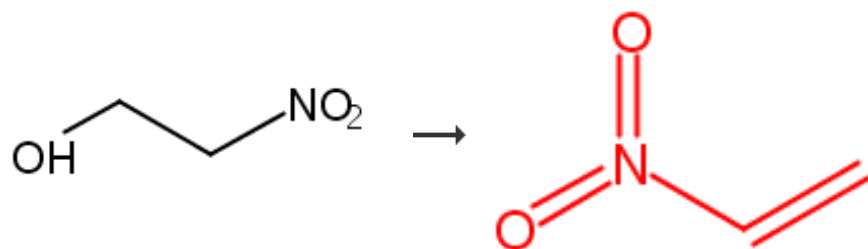
## Available Experimental Data

<sup>1</sup>H NMR

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## 20. Single Step



### Overview

#### Steps/Stages

1.1 R:EtN(Pr-)<sub>2</sub>, R:MeSO<sub>2</sub>Cl, S:PhMe

#### Notes

Reactants: 1, Reagents: 2, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[Asymmetric Organocatalysis in Flow](#)

By Jin, Xiaoliang

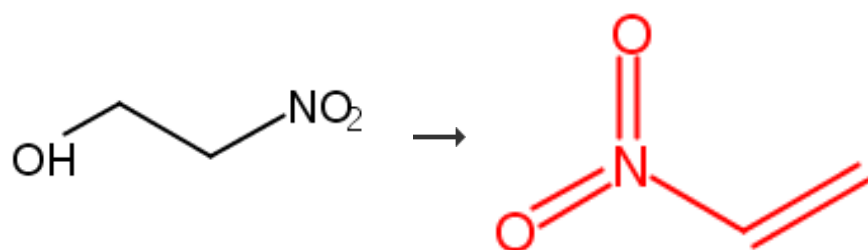
From null, , No pp.; 2010

### Experimental Procedure

Preparation of nitroethylene: B) Procedure in flow: A stock solution of 2-nitroethanol (35.7 microL, 0.5 mmol) and mesyl chloride (38.6 microL, 0.5 mmol) in toluene (1 mL) was prepared and injected into one of the sample loops; Another stock solution of Hunig's base (175 microL, 1 mmol) in toluene (1 mL) was also prepared and injected into a separate sample loops of the R2+ unit of the Vapourtec system. The resulting streams were mixed in a T-piece and then directly into a flow coil (10 mL volume, 1.0 mm i.d.) mounted on the R4 unit to give residence time of 5 min with a total flow rate of 2 mL/min. After leaving the coil the reaction mixture was directed into a glass column containing the scavenging resin A15 (polymer-supported acid) (1.1 g, 1.1 mmol). The purified product 50 was then used directly for the Michael addition reaction.

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#### 21. Single Step



### Overview

#### Steps/Stages

1.1 R:Phthalic anhydride

#### Notes

Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[Asymmetric Organocatalysis in Flow](#)

By Jin, Xiaoliang

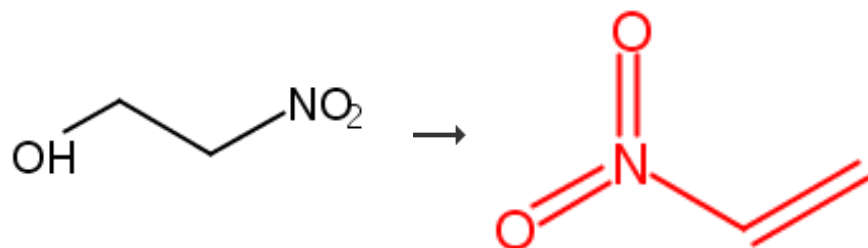
From null, , No pp.; 2010

### Experimental Procedure

Preparation of nitroethylene: A) Procedure in batch: 2-Nitroethanol (0.36 mL, 5 mmol) and phthalic anhydride (0.96 g, 6.5 mmol) were mixed in a 50 mL round bottom flask equipped with a magnetic stir bar. The flask was then equipped with a vacuum distillation setup with a fractional distillation column and a -78 degC bath-cooled receiver. The apparatus was evacuated to 60 mm Hg, and heated to and maintained at 140 degC. The starting materials turned into a homogeneous solution, and the distillate was collected until the distillation ceased to give a pale yellow solid at -78 degC. The solid, containing a mixture of nitroethylene and water, was warmed in an ice-water bath to give a pale yellow heterogeneous mixture. The mixture was mixed with toluene, dried over anhydrous CaCl<sub>2</sub> and filtered through a pad of anhydrous CaCl<sub>2</sub>. The filtrate was collected as a pale yellow stock solution of nitroethylene in toluene, and stored at -20 degC for future use. The concentration of nitroethylene of the stock solution can be estimated via NMR analysis in chloroform (with toluene as internal standard). Nitroethylene as a solution in dry toluene was found to be stable (no change in NMR) for at least 3 months when stored in a freezer.

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## 22. Single Step



### Overview

#### Steps/Stages

1.1

#### Notes

no exptl. detail, literature prepn., Reactants: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

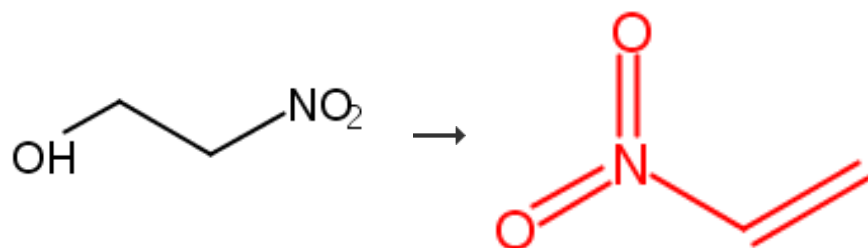
[Dioxindole in asymmetric catalytic synthesis: direct access to 3-substituted 3-hydroxy-2-oxindoles via 1,4-additions to nitroalkenes](#)

By Retini, Michele et al

From *Chemical Communications* (Cambridge, United Kingdom), 48(27), 3336-3338; 2012

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## 23. Single Step





[Overview](#)**Steps/Stages**

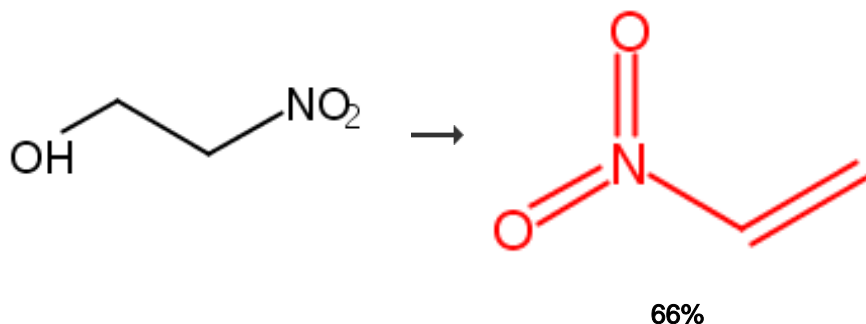
1.1 C:Phthalic anhydride, 140-180°C, 80 mmHg

**Notes**Reactants: 1, Catalysts: 1, Steps: 1, Stages: 1,  
Most stages in any one step: 1**References**[Nitroethylene](#)

By Singleton, Daniel A.

From e-EROS Encyclopedia of Reagents for  
Organic Synthesis, , No pp. given; 2001

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**24. Single Step**[Overview](#)**Steps/Stages**

1.1

**Notes**

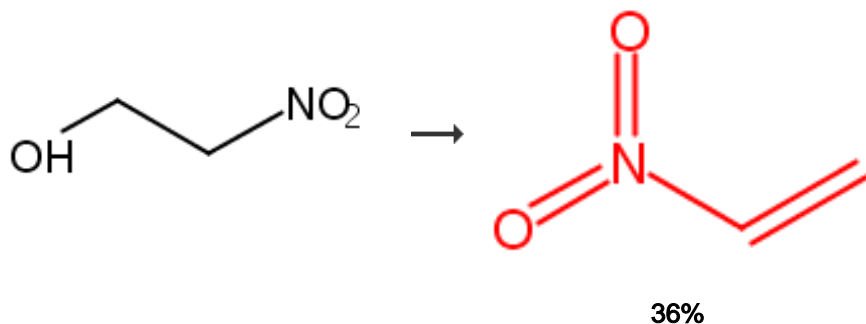
Classification: Elimination; Dehydration; #  
 Conditions: phthalic anhydride; 140-150 deg /80mm; 175-180 deg vac distil, Reactants: 1,  
 Steps: 1, Stages: 1, Most stages in any one step: 1

**References**[Aliphatic nitro compounds. I. Preparation of nitro olefins by dehydration of 2-nitro alcohols](#)

By Buckley, G. D. and Scaife, C. W.

From Journal of the Chemical Society, , 1471-2; 1947

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**25. Single Step**

[Overview](#)**Steps/Stages**

1.1 R:Phthalic anhydride

**Notes**

Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

**References**[Asymmetric synthesis of aphanorphine and synthetic approaches towards dietyoxetane](#)

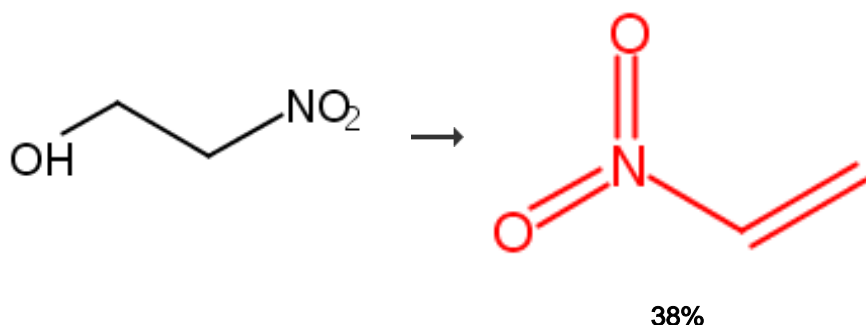
By Welsh, Emma Jane

From null, , No pp.; 2007

[Experimental Procedure](#)

Nitroethene 534: A mixture of nitroethanol 533 (518 was author's error) (18.96 g, 208 mmol) and phthalic anhydride (34.25 g, 231 mmol) was heated at reflux at 180 degC for 45 min. The mixture was allowed to cool to rt before replacing the condenser with a distillation apparatus. Nitroethene was distilled from the crude reaction mixture under vacuum into a receiving flask containing CaCl<sub>2</sub> and then redistilled from CaCl<sub>2</sub> to provide the title compound (5.50 g, 36 %). (LACHRYMATOR!) Nitroethene was stored over CaCl<sub>2</sub> at -18 degC. bp 97-99 degC.

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**26. Single Step**[Overview](#)**Steps/Stages**

1.1 R:Phthalic anhydride

**Notes**

Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

**References**[Studies towards the total synthesis of \(-\)mitragynine using solid-supported reagents](#)

By Henry, D. J.

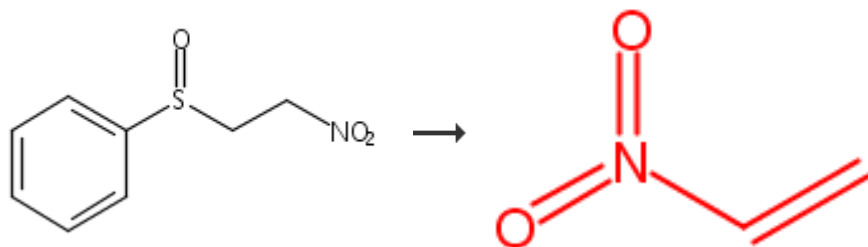
From null, , No pp.; 2003

[Experimental Procedure](#)

Nitroethylene 86: 2-Nitroethanol 90 (3.96 g, 43.5 mmol) was added to phthalic anhydride (9.7 g, 65 mmol) and the mixture heated to 150 degC at approx. 80 mm Hg until the mixture had homogenised. The product was then distilled at 180 degC, and the resultant yellow liquid was dried over calcium chloride and filtered to afford nitroethylene 86 (1.22 g, 38 %).

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**27. Single Step**



### Overview

#### Steps/Stages

1.1 S:Benzene

#### Notes

Reactants: 1, Solvents: 1, Steps: 1, Stages: 1,  
Most stages in any one step: 1

#### References

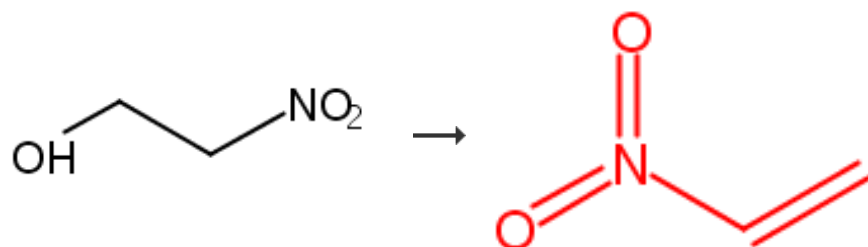
[2-Nitroethyl phenyl sulfoxide. A novel reagent for facile nitroethylene transfer](#)

By Ranganathan, Darshan et al

From Journal of Chemical Research, Synopses, (3), 78-9; 1983

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#### 28. Single Step



### Overview

#### Steps/Stages

1.1 C:Phthalic anhydride

#### Notes

Reactants: 1, Catalysts: 1, Steps: 1, Stages: 1,  
Most stages in any one step: 1

#### References

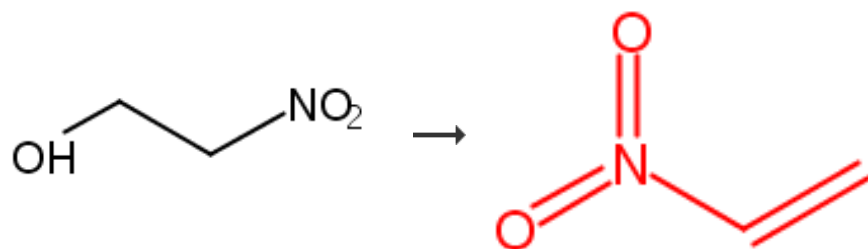
[Nitroethylene: a stable, clean, and reactive agent for organic synthesis](#)

By Ranganathan, Darshan et al

From Journal of Organic Chemistry, 45(7), 1185-9; 1980

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#### 29. Single Step



### Overview

#### Steps/Stages

1.1

#### Notes

Classification: Elimination; Dehydration; #  
 Conditions: phthalic anhydride, Reactants: 1,  
 Steps: 1, Stages: 1, Most stages in any one  
 step: 1

#### References

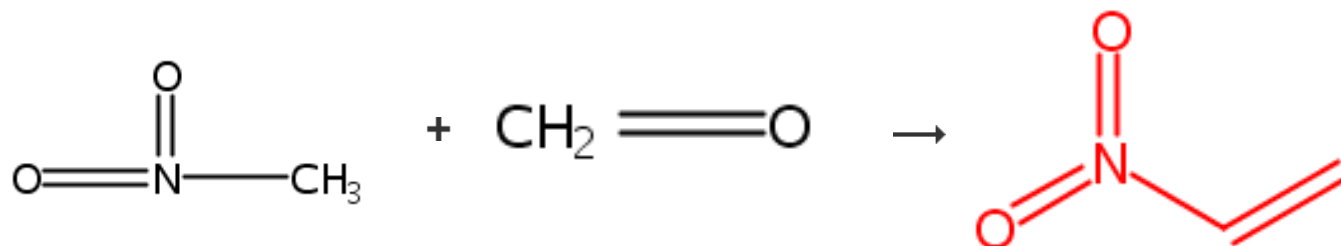
[Polycyclic compounds containing nitrogen. III. Diels-Alder reaction of nitroethylene](#)

By Drake, Nathan L. and Kraebel, Charlotte M.

From Journal of Organic Chemistry, 26, 41-5; 1961

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#### 30. 2 Steps



### Overview

#### Steps/Stages

1.1 R:KOH, S:MeOH

2.1 R:Phthalic anhydride

#### Notes

Reactants: 2, Reagents: 2, Solvents: 1, Steps: 2, Stages: 2, Most stages in any one step: 1

#### References

[Asymmetric synthesis of aphanorphine and synthetic approaches towards dietyoxetane](#)

By Welsh, Emma Jane

From null, , No pp.; 2007

### Experimental Procedure

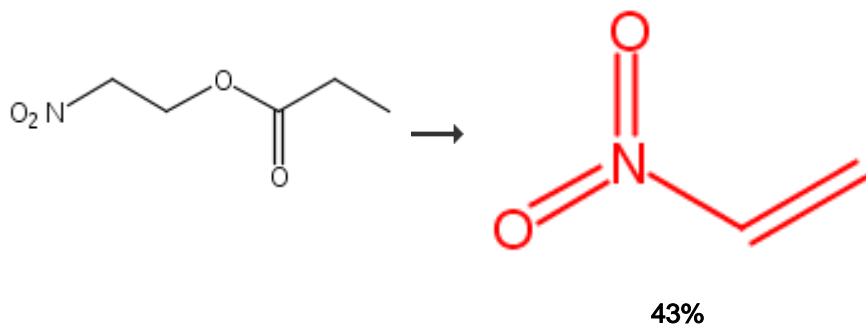
**Step 1**

The crude nitroethanol 533 (518 was author's error): A three necked round bottomed flask was equipped with a dropping funnel, thermometer and a condenser. Paraformaldehyde (18.0 g, 0.6 mol) and freshly distilled nitromethane (340 mL, 6.7 mol) were added to the flask and the mixture was stirred vigorously. KOH (3 M in MeOH, Ca. 1.5 mL) was added dropwise through the dropping funnel until pH 8 was reached (litmus paper). The mixture became homogenous after 20 min and the solution was stirred at rt for 1 hr. Conc. H<sub>2</sub>SO<sub>4</sub> was added until the solution reached pH 4 and the solution was stirred at rt overnight. The reaction mixture was subjected to vacuum filtration and the filtrate (66 g) was added to a flask containing diphenyl ether (66 g). The solution was distilled under reduced pressure (0.8 mbar). The first fraction to distil contained unreacted nitromethane (bp 24 degC), the second fraction contained nitroethanol and diphenyl ether (bp 84 degC). The second fraction was biphasic and the lower layer was removed and washed with hexane (30 mL). The lower layer was separated to afford nitroethanol (23.88 g, 44 %) and was used without further purification.

**Step 2**

Nitroethene 534: A mixture of nitroethanol 533 (518 was author's error) (18.96 g, 208 mmol) and phthalic anhydride (34.25 g, 231 mmol) was heated at reflux at 180 degC for 45 min. The mixture was allowed to cool to rt before replacing the condenser with a distillation apparatus. Nitroethene was distilled from the crude reaction mixture under vacuum into a receiving flask containing CaCl<sub>2</sub> and then redistilled from CaCl<sub>2</sub> to provide the title compound (5.50 g, 36 %). (LACHRYMATOR!) Nitroethene was stored over CaCl<sub>2</sub> at -18 degC. bp 97-99 degC.

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**31. Single Step**[Overview](#)**Steps/Stages**

1.1 R:CaSO<sub>4</sub>

**Notes**

Classification: Elimination; # Conditions: CaSO<sub>4</sub> heat; 240-260 deg; Reactants: 1, Reagents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

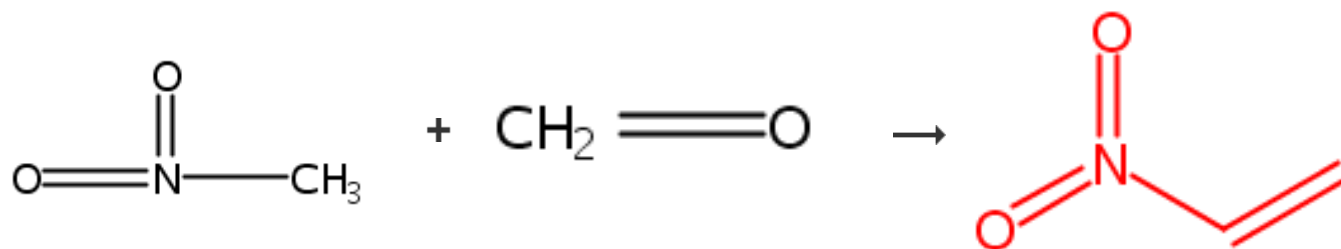
**References**[Nitroethylene](#)

By Hopff, H. and Capaul, M.

From Helvetica Chimica Acta, 43, 1898-910; 1960

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**32. 2 Steps**



### Overview

#### Steps/Stages

1.1 S:H<sub>2</sub>O

2.1 R:Phthalic anhydride

#### Notes

Reactants: 2, Reagents: 1, Solvents: 1, Steps: 2, Stages: 2, Most stages in any one step: 1

#### References

[Studies towards the total synthesis of \(-\)-mitragynine using solid-supported reagents](#)

By Henry, D. J.

From null, , No pp.; 2003

### Experimental Procedure

#### Step 1

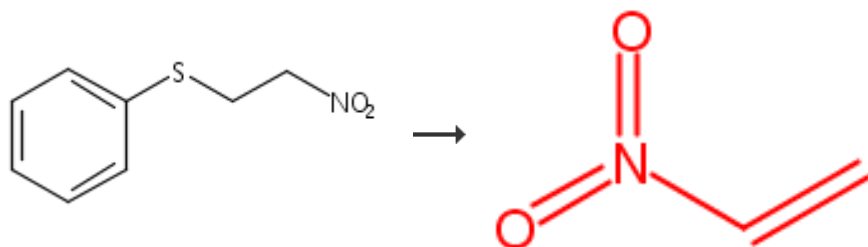
2-Nitroethanol 90: Formaldehyde (67 % aqueous solution) (10 mL, 193 mmol) was added to a suspension of polymer-supported hydroxide (95 g, 285 mmol) in nitromethane (200 mL) and the reaction mixture stirred at rt for 10 min. The mixture was filtered, the beads washed extensively with nitromethane (3 x 100 mL) and CH<sub>2</sub>Cl<sub>2</sub> (5 x 100 mL) and the solvent then removed in vacuo. The resultant yellow oil was redissolved in CH<sub>2</sub>Cl<sub>2</sub> (100 mL), dried (MgSO<sub>4</sub>) and the solvent removed in vacuo to afford 2-nitroethanol 90 as a yellow oil (3.96 g, 23 %).

#### Step 2

Nitroethylene 86: 2-Nitroethanol 90 (3.96 g, 43.5 mmol) was added to phthalic anhydride (9.7 g, 65 mmol) and the mixture heated to 150 degC at approx. 80 mm Hg until the mixture had homogenised. The product was then distilled at 180 degC, and the resultant yellow liquid was dried over calcium chloride and filtered to afford nitroethylene 86 (1.22 g, 38 %).

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### 33. 2 Steps

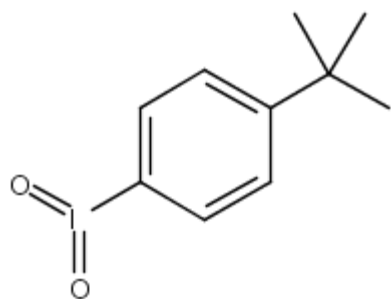


### Overview

#### Steps/Stages

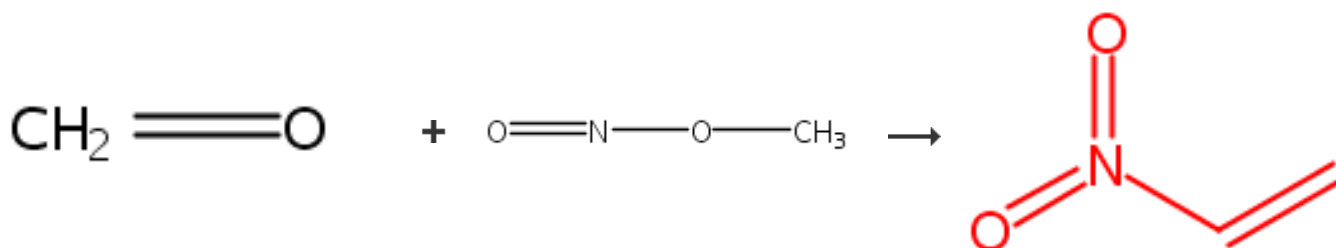
#### Notes

1.1 R:

R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

2.1 S:Benzene

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**34. 2 Steps****Overview****Steps/Stages**

1.1  
2.1 C:Phthalic anhydride

**Notes**

Reactants: 2, Catalysts: 1, Steps: 2, Stages: 2,  
Most stages in any one step: 1

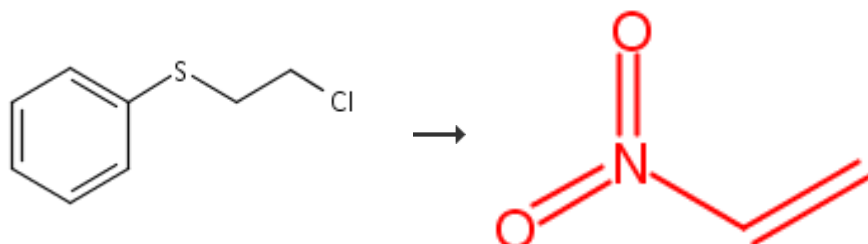
**References**

[Nitroethylene: a stable, clean, and reactive agent for organic synthesis](#)

By Ranganathan, Darshan et al

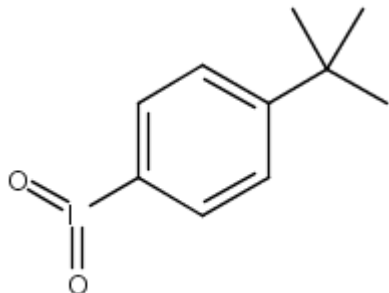
From Journal of Organic Chemistry, 45(7), 1185-9; 1980

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**35. 3 Steps**

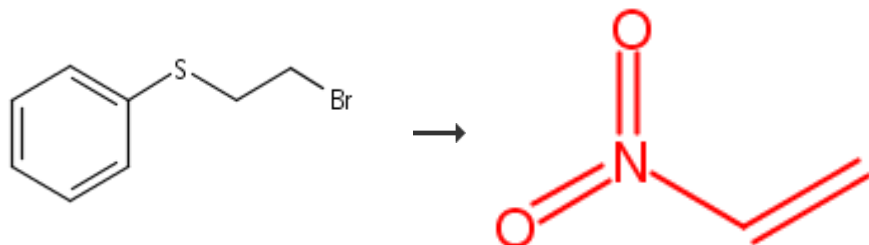
[Overview](#)**Steps/Stages**1.1 R:AgNO<sub>2</sub>, S:Et<sub>2</sub>O

2.1 R:

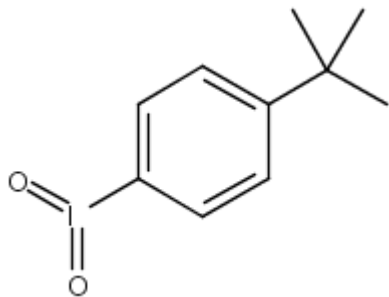
R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

3.1 S:Benzene

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**36. 3 Steps**[Overview](#)**Steps/Stages**1.1 R:NaNO<sub>2</sub>, S:DMSO

2.1 R:

R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

3.1 S:Benzene

**Notes**

Reactants: 1, Reagents: 3, Solvents: 2, Steps: 3, Stages: 3, Most stages in any one step: 1

**References**

[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987

**Notes**

Reactants: 1, Reagents: 3, Solvents: 2, Steps: 3, Stages: 3, Most stages in any one step: 1

**References**

[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

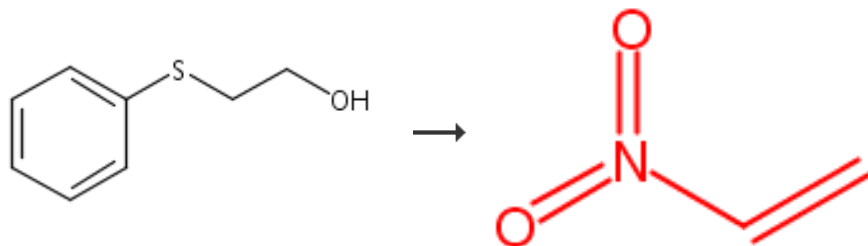
By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987



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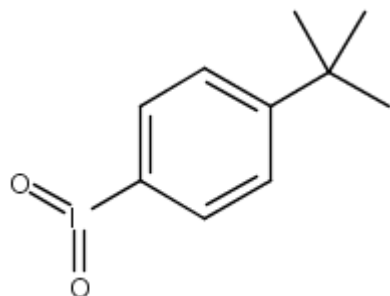
### 37. 4 Steps



#### Overview

#### Steps/Stages

- 1.1 R:SOCl<sub>2</sub>
- 2.1 R:AgNO<sub>2</sub>, S:Et<sub>2</sub>O
- 3.1 R:

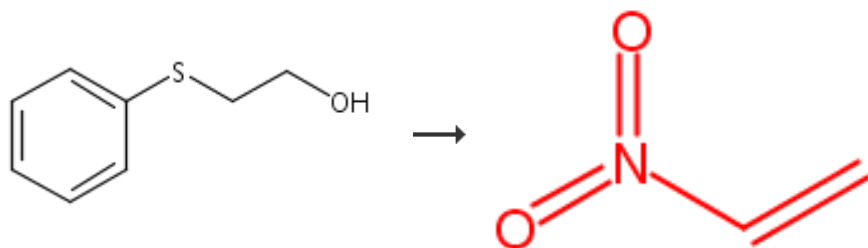


R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

- 4.1 S:Benzene

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### 38. 4 Steps



#### Overview

#### Steps/Stages

#### Notes

Reactants: 1, Reagents: 4, Solvents: 2, Steps: 4, Stages: 4, Most stages in any one step: 1

#### References

[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

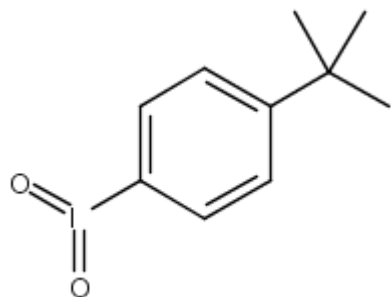
By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987

#### Notes

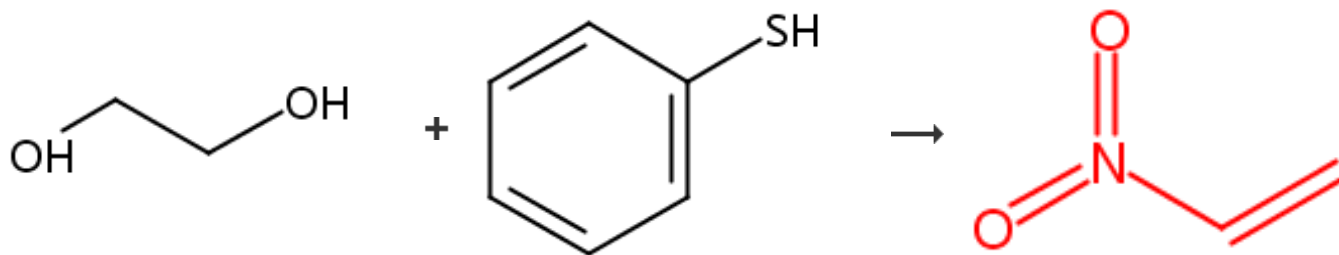
1.1 R:PBr<sub>3</sub>2.1 R:NaNO<sub>2</sub>, S:DMSO

3.1 R:

R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

4.1 S:Benzene

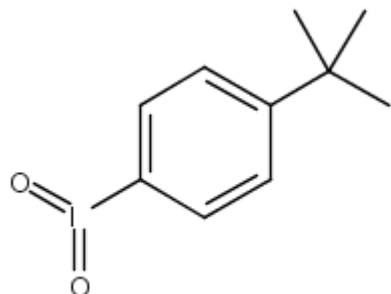
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**39. 6 Steps**

[Step 2.1]

**Overview****Steps/Stages**1.1 R:HBr, S:H<sub>2</sub>O2.1 R:NaOH, S:H<sub>2</sub>O3.1 R:SOCl<sub>2</sub>4.1 R:AgNO<sub>2</sub>, S:Et<sub>2</sub>O

5.1 R:

R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

6.1 S:Benzene

Reactants: 1, Reagents: 4, Solvents: 2, Steps: 4, Stages: 4, Most stages in any one step: 1

**References**

[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987

**Notes**

Reactants: 2, Reagents: 6, Solvents: 3, Steps: 6, Stages: 6, Most stages in any one step: 1

**References**

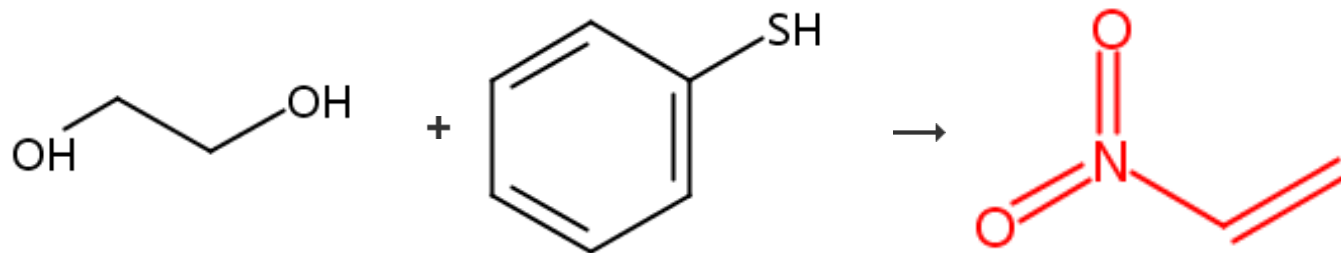
[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987

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#### 40. 6 Steps

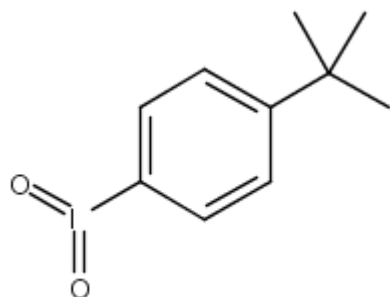


[Step 2.1]

#### Overview

##### Steps/Stages

- 1.1 R:HBr, S:H<sub>2</sub>O
- 2.1 R:NaOH, S:H<sub>2</sub>O
- 3.1 R:PBr<sub>3</sub>
- 4.1 R:NaNO<sub>2</sub>, S:DMSO
- 5.1 R:

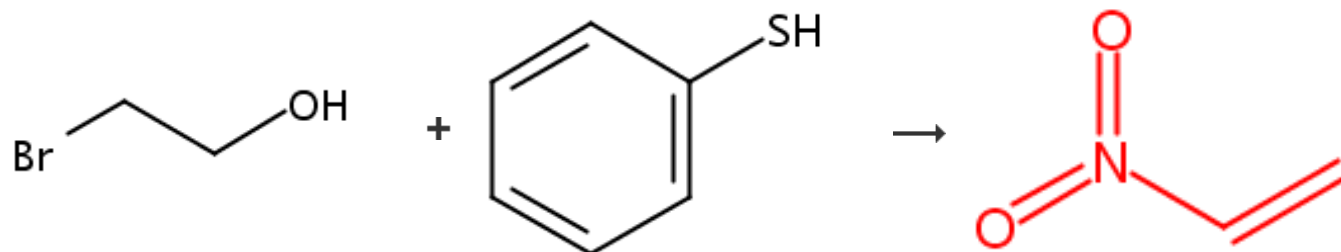


R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

- 6.1 S:Benzene

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#### 41. 5 Steps



#### Overview

##### Notes

Reactants: 2, Reagents: 6, Solvents: 3, Steps: 6, Stages: 6, Most stages in any one step: 1

##### References

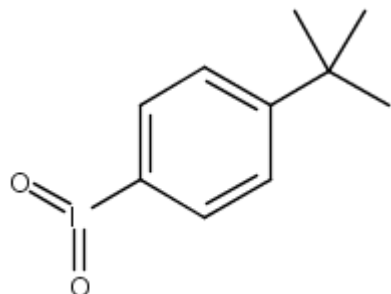
[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987

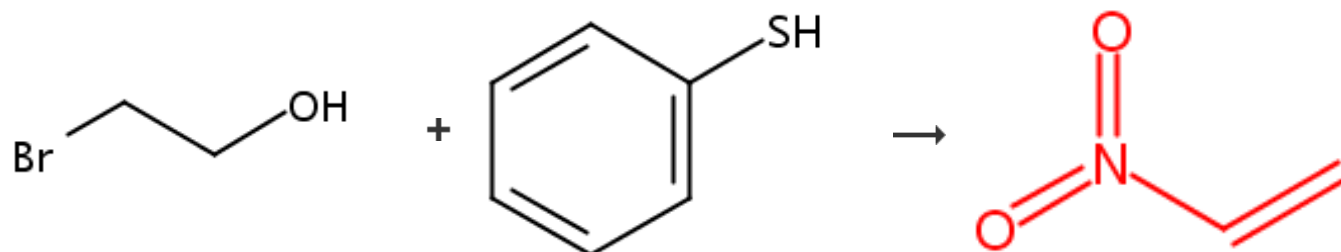
**Steps/Stages**1.1 R:NaOH, S:H<sub>2</sub>O2.1 R:SOCl<sub>2</sub>3.1 R:AgNO<sub>2</sub>, S:Et<sub>2</sub>O

4.1 R:

R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

5.1 S:Benzene

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**42. 5 Steps**[Overview](#)**Steps/Stages****Notes**

Reactants: 2, Reagents: 5, Solvents: 3, Steps: 5, Stages: 5, Most stages in any one step: 1

**References**

[A practical and convenient synthesis of the nitroethylene transfer reagent, 2-nitroethyl phenyl sulfoxide](#)

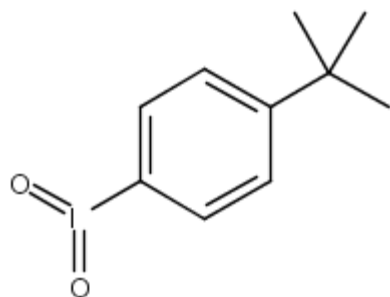
By Ranganathan, S. et al

From Tetrahedron Letters, 28(25), 2893-4; 1987

**Notes**

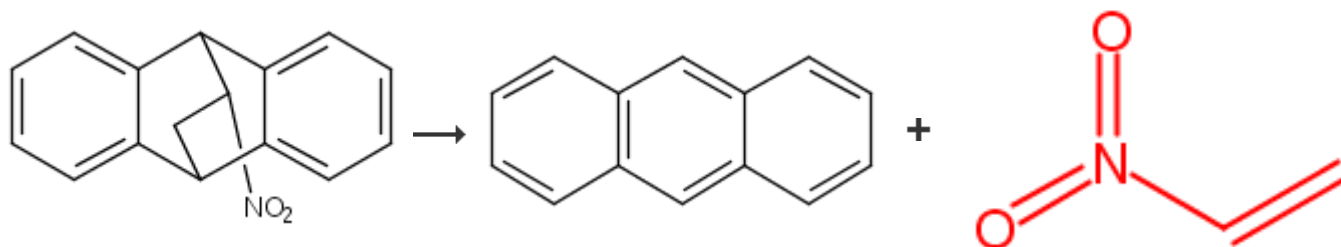
1.1 R:NaOH, S:H<sub>2</sub>O2.1 R:PBr<sub>3</sub>3.1 R:NaNO<sub>2</sub>, S:DMSO

4.1 R:

R:F<sub>3</sub>CCO<sub>2</sub>H, S:Benzene

5.1 S:Benzene

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**43. Single Step****Overview****Steps/Stages**

1.1 250°C

**Notes**

Aromatization, Cleavage, Ring cleavage, thermal, Reactants: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

**References**

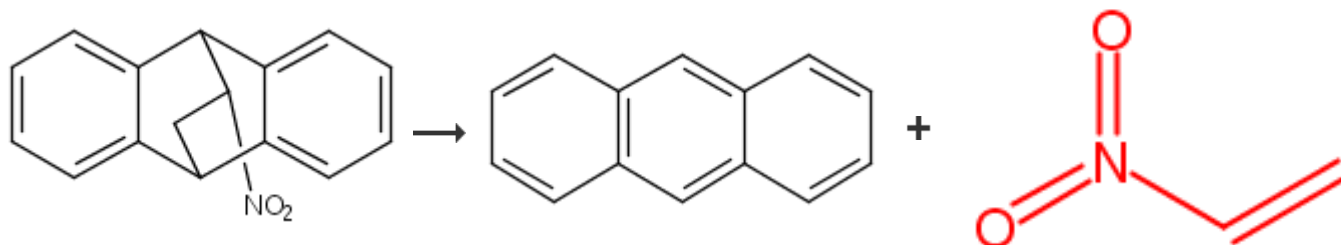
[The retro-Diels-Alder reaction. Part I. C-C dienophiles](#)

By Rickborn, Bruce

From Organic Reactions (Hoboken, NJ, United States), 52, No pp. given; 1998

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**44. Single Step**



### Overview

#### Steps/Stages

1.1 250°C

#### Notes

Cleavage, Ring cleavage, thermal, Reactants: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

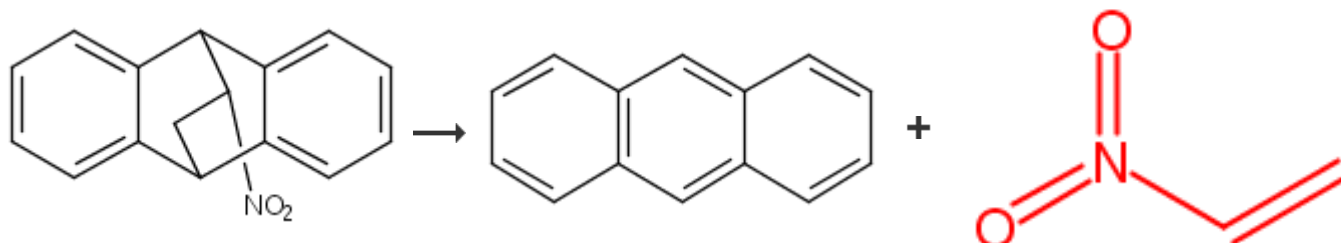
[The retro-Diels-Alder reaction. Part I. C-C dienophiles](#)

By Rickborn, Bruce

From Organic Reactions (Hoboken, NJ, United States), 52, No pp. given; 1998

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#### 45. Single Step



### Overview

#### Steps/Stages

1.1 S:PhOPh, 250°C

#### Notes

Cleavage, Ring cleavage, thermal, Reactants: 1, Solvents: 1, Steps: 1, Stages: 1, Most stages in any one step: 1

#### References

[The retro-Diels-Alder reaction. Part I. C-C dienophiles](#)

By Rickborn, Bruce

From Organic Reactions (Hoboken, NJ, United States), 52, No pp. given; 1998

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